

balchem[™]

Metalosate[™]

Palm Oil Metalosate Boron

Roberto Gonzalez

Metalosate

Amino Acid Chelate Liquid Foliar Fertilizers



Nutrition Delivered™

Oil Palm is one of the 16 Crops most sensitive to Boron deficiency



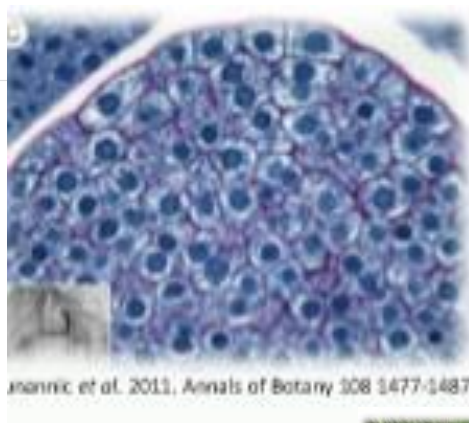
Deficiency leaf and Fruit in Palm Oil

Boron functions

It is part of the cell wall that confers tissue rigidity and resistance to mechanical stress.
B is part of the cell membrane

-B thick cell wall, brittle altered mechanical properties does not expand properly Al accumulation points are generated in roots

-B lower absorption of P in radical leaking points of K, sugars, amino acids and phenols

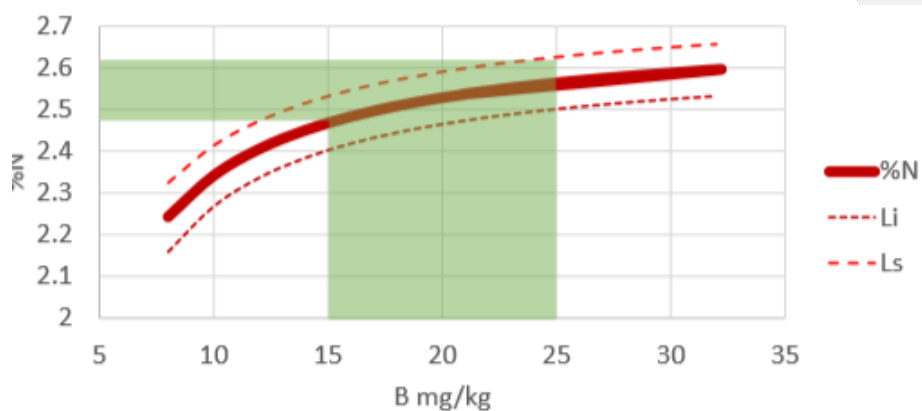


Cell Wall and Plasma Membrane

Boron functions

N metabolism (nitrate reductase)

-B has rapid and profound effects, it triggers changes that are assumed to be $f B$:
 Transport of sugars
 Metabolism of phenols



Ratio model [% N foliar] - [B mg / kg] in sheet 17, based on 500 commercial planting records 2012-13, AIPA, Colombia (2015-2018)

Plant metabolism

Boron functions

-B viability and germination of pollen and elongation of the pollen tube Stigma must provide the B ... it is limited by less transpiration Before symptoms of foliage deficiency it is probable that the pollination or fertilization is deficient



Reproductive growth

Boron functions

sprouts
-B Cessation of elongation and meristematic growth of outbreaks increased AIA is a side effect of - B but is not the one that stops growth



Root elongation and shoot growth

Absorption of boron by plants

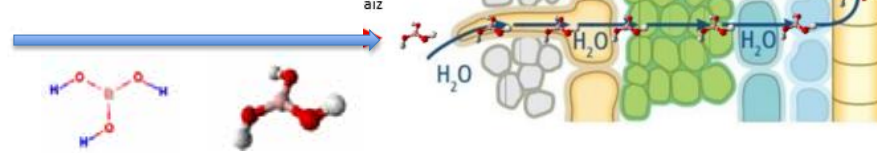
The dissociation of borates in the soil is determined by:

- pH gives solution
- Organic material
- Types of clay
- Humidity of floor
- Soil temperature

These variables determine the amount of B that will be dissociated in boric acid and that will be available in the soil solution for plant absorption.



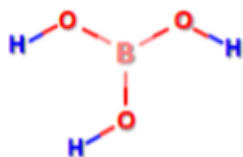
Plants absorb boron from the soil in the form of dissociated boric acid¹



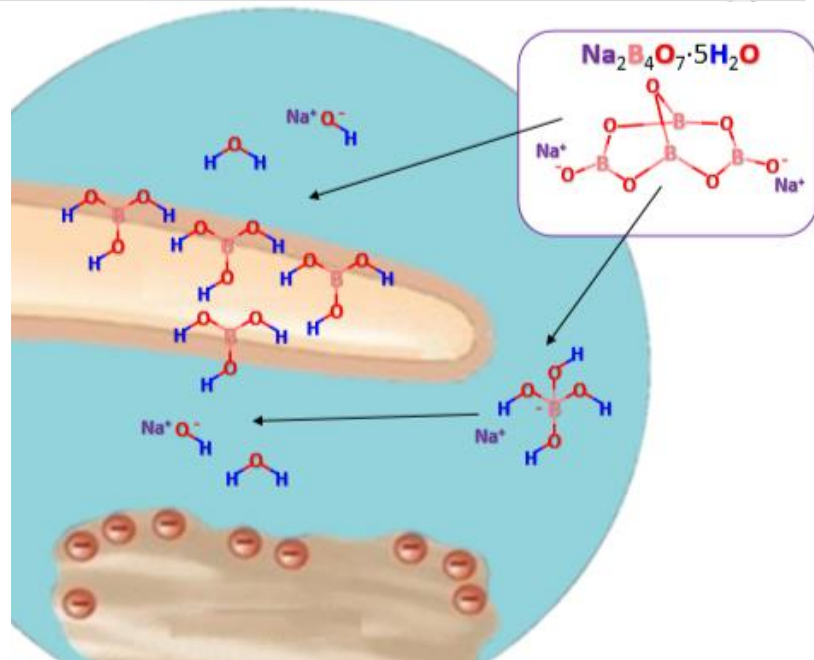
1. Brown, P. and Hill, H. 1997. Absorption of boron by plant roots. Plant and Soil 193: 49-56.
 2. Goldberg, S., 1997 Reactions of boron with soils. Plant and Soil 193: 35-48.

Dissociation of sodium tetraborate in the soil solution

Plants absorb boron in the form of $B(OH)_3$
 Dissociation of sodium tetraborate in the soil solution



Four B molecules are released into the soil solution as boric acid, which is bioavailable for absorption by plants along with water.



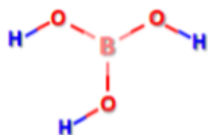
Absorption Boron

Dissociation of sodium and calcium pentaborate in the soil solution

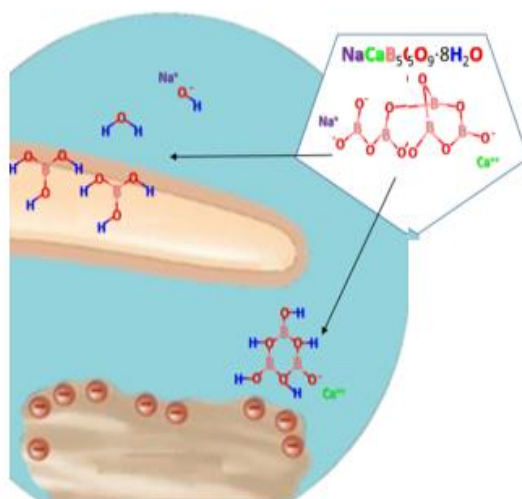
Plants absorb boron in the form of B(OH)₃

Dissociation of sodium and calcium pentaborate in the soil solution

$\text{NaCaB}_5\text{O}_9 \cdot 8\text{H}_2\text{O}$



Only two B molecules are released into the soil solution as boric acid, which is bioavailable for absorption by plants along with water



Three B molecules become strongly linked to the Ca ++ molecule, forming a calcium pentahydroxyborate. The boron contained in this molecule will not be bioavailable for absorption by plants for a very long period.

Absorption Boron

Conditions Associated With Boron Deficiency

Soil

- B adsorbs strongly to amorphous Fe and Al oxides, Mg hydroxide and MO
- B that is not adsorbed is highly susceptible to leaching
- B-poor soil (<0.5 mg / l):
- coarse textures (sandy)
- low MO (<2.5%) ~ controlled release of B • organic soils (peat) • very old (no deposition of rejuvenating materials such as ash or alluvium)
- high fertility or fertilization with N, K, Ca • 4.5 <pH> 7.5

Crops

- Highly productive and nutrient-demanding materials (N, P, K)
- Diseases that affect the radical system
- Little perspiration ...

Weather

- Water deficit
- Heavy, abundant or irregularly distributed precipitation

Palm Oil deficiency Boron

Soil pH affects the absorption of Boron

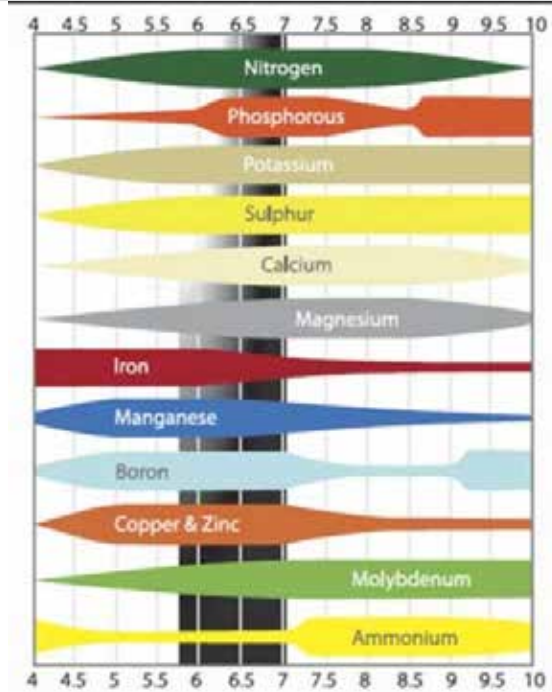


Table pH

Peat Soil

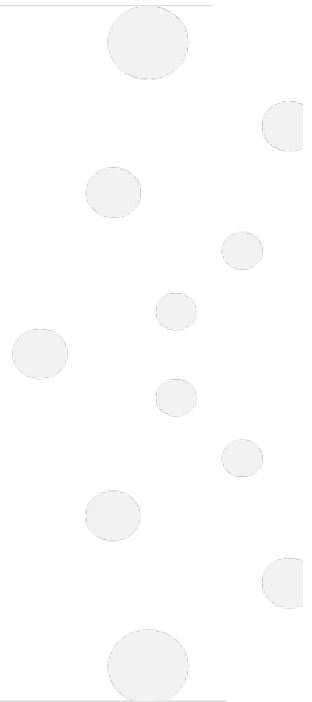
The pH is very acidic that does not allow the absorption of microelements

The high content of organic matter sequesters the microelements



Picture Indonesia

Peat Soil



Sandy Soil

Absorption of nutrients is limited and easily leached



Boron Deficiency



Curly Leaf

Boron Deficiency



Blind Leaf

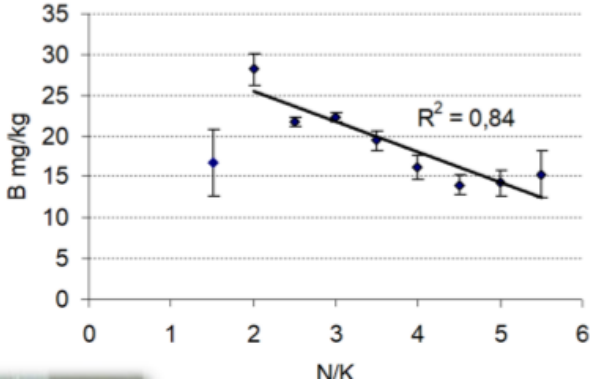
Boron Deficiency



White Strip Leaf

White Strip Leaf

Unbalance N/K-B



Boron Deficiency

Boron Deficiency



Hook leaf

Boron Deficiency



Bonefish Leaf

Boron deficiency leads to decreased pollination

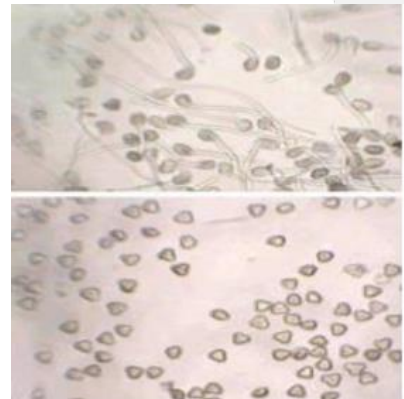


Anthesis period 1 - 5 days

Pollen Viability: Boron deficiency reduces pollen germination and male sterility. Oil palm - reduced viability due to weather



Anthesis Period 1 - 3 days The highest pollination activity occurs on the day



Sprouted and non-sprouted pollen

Dr Ramles Moslin

Boron deficiency produces a low pollination of the Fruits



Good pollination 70% or more Fruit



Medium Pollination 30%- 70% Fruit



Low Pollination 30% or less Fruit

Dr Ramles Moslin

Boron deficiency in fresh fruits leads to parthenocarpic fruits (seedless), small fruits



Boron Deficiency

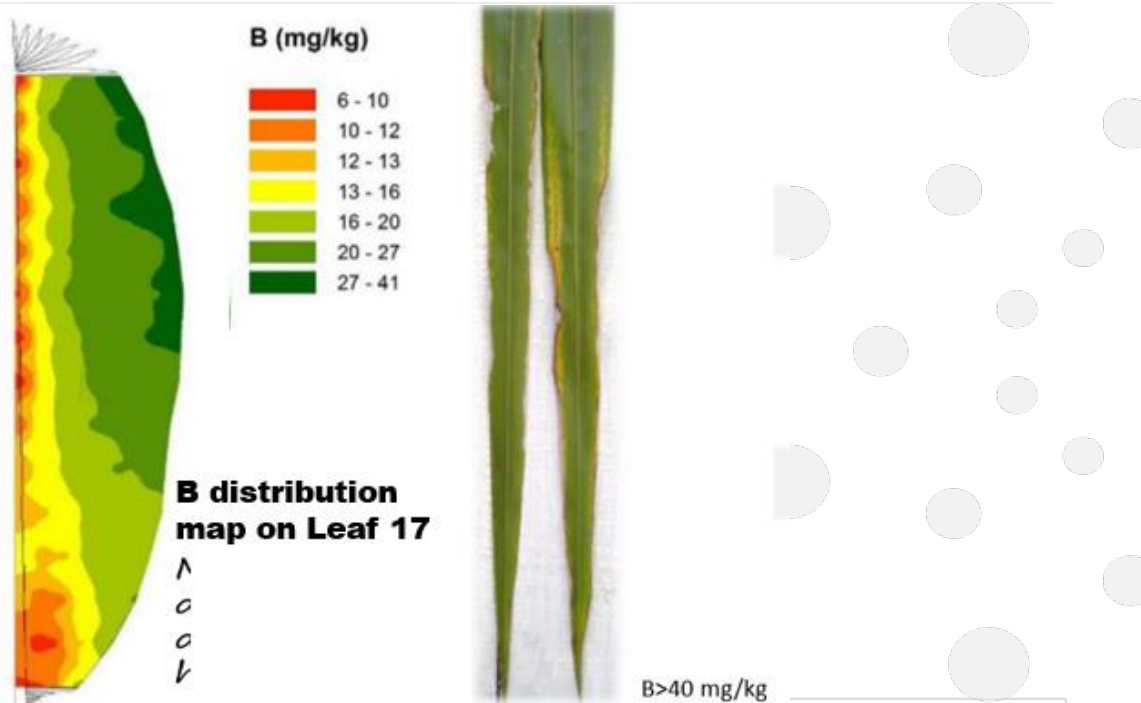
Cause of Boron Toxicity



- Concentrated application
- Armpit application: it is associated with a rapid absorption of B but increases the risk of toxicity
- Application of B in summer: in the soil the B is concentrated (OM, restitution) and with the first rains the availability can be very high
- In nursery: application of B with limited irrigation
- Irrigation water with high [B]

Boron Toxicity

Symptoms of Toxicity



Toxicity Boron

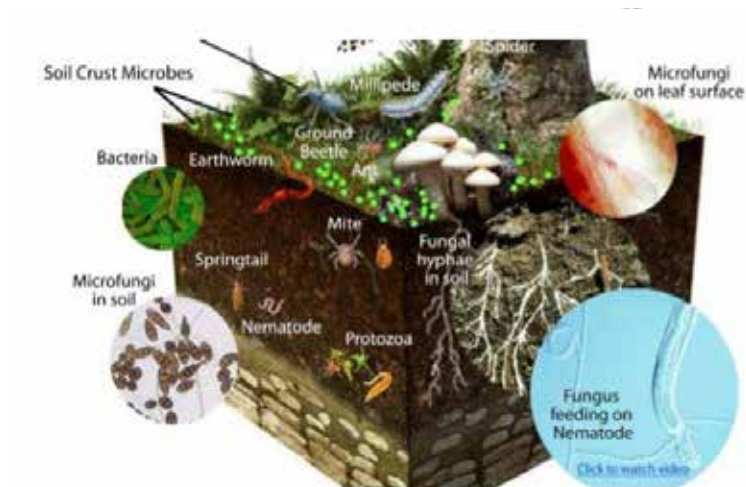
Boron Toxicity



Boron toxicity in nursery

Biocide in high concentrations Boron

"All things are poisonous and nothing is without poison, it is the dose that makes the poison". Paracelsus



Die dritte Defension wezen des Schreiber der neuen Rezepte. "Septem Defensiones 1538. Werke Bd. 2. Darmstadt 1965. p. 510

Toxic Boron High concentration in Soil

Effect toxic high Concentration Boron

Em 2010, Leonie Becker, Adam Scheffczyk, Bernhard Förster, Jörg Oehlmann, Juliska Princz, Jörg Römbke e Thomas Moser do Instituto Soil Science and Soil Conservation, Research Centre for BioSystems, Land Use and Nutrition de Alemanha avaliaron o efeito do acido borico em varios macro e microorganismos

Test organism	Endpoint	EC ₅₀ (l/u 95% cl) [mg/kg soil (dw)]
Nitrogen transformation	Nitrate formation	>2,400
<i>Avena sativa</i>	Biomass fw; Shoot length	182 (119/281); 308 (n. d.)
<i>Brassica napus</i>	Biomass fw; Shoot length	175 (164/187); 357 (354/361)
<i>Caenorhabditis elegans</i>	Reproduction	747 (589/949)
<i>Enchytraeus crypticus</i>	Reproduction	220 (208/233)
<i>Enchytraeus luxuriosus</i>	Reproduction	228 (201/259)
<i>Eisenia fetida</i>	Reproduction	484 (465/504)
<i>Folsomia candida</i>	Avoidance	1,441 (183/478,293)
<i>Poecilus cupreus</i>	Food uptake	1,342 (249/27,465)

Amount of information on the toxicity of boric acid to soil organisms (including microbes and plants) is sufficient to support the idea of using this chemical as a reference substance in terrestrial Eco toxicological tests.

Toxicity Boron

What are the effects of boron on organisms in the environment?

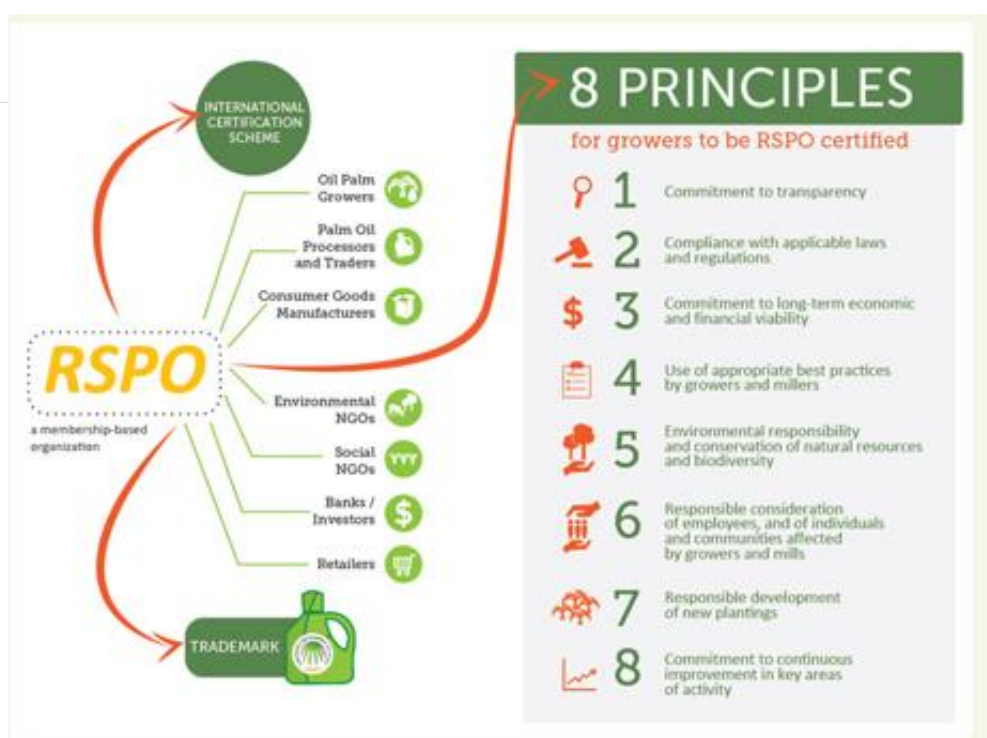
Table: "Reported critical boron levels for several types of environmental organism"

Type of organism	Type of effect(s) reported	Boron concentration in the water
Bacteria	Mixture of acute and chronic effects, for several types of bacteria	0 - 340 mg/litre (mainly above 18 mg/litre)
Protozoa	First very small effects, for 2 types of protozoa	0.3 - 18 mg/litre
Freshwater green algae	Highest concentration with no effect (NOEC), for several types of algae	10 - 24 mg/litre
Blue-green algae	First very small effect for 1 type of blue-green algae	20 mg/litre
Invertebrates	Acute effects for several types of invertebrates	95 to 1376 mg/litre (mainly 100-200 mg/litre)
Water flea (Daphnia magna)	Highest concentration with no effect (NOEC), found in many chronic tests for this one invertebrate type	6 - 10 mg/litre
Fish	Acute effects in several types of fish	10 to nearly 300 mg/litre
Rainbow trout	Chronic tests with standard laboratory water (NOEC)	0.009 - 0.103 mg/litre
Rainbow trout	Chronic tests in several natural waters (NOEC)	0.75 - > 18 mg/litre

The data in the table show that bacteria are much less sensitive to boron, compared to other chemicals. Protozoa are somewhat more sensitive. Algae, for which boron is an essential nutrient, also have low sensitivity to boron. Invertebrates also have a low boron sensitivity, as determined from many long-term studies. Fish are the most sensitive species to boron.

PROTECTION, CONSERVATION AND IMPROVEMENT OF ECOSYSTEMS AND THE ENVIRONMENT





The following **eight principles** must be observed by RSCO members:

Boron Management Problems

Ensuring an adequate supply and absorption of B throughout the year constitutes the main challenge for the management of B in oil palm

1- Leaf chemical analysis of the leaf (1, 3, 9, 17): allows monitoring [B] on the sampling leaf shows the “accumulated history” on the leaf does not indicate the status of B in sensitive tissues (meristems) Symptoms of deficiency of B in leaf 1, reveal the deficit in the supply of B until 5 months ago, when the now visible tissue was in the bud

2- Water deficit there may be high [B] in the soil, but there is no absorption.

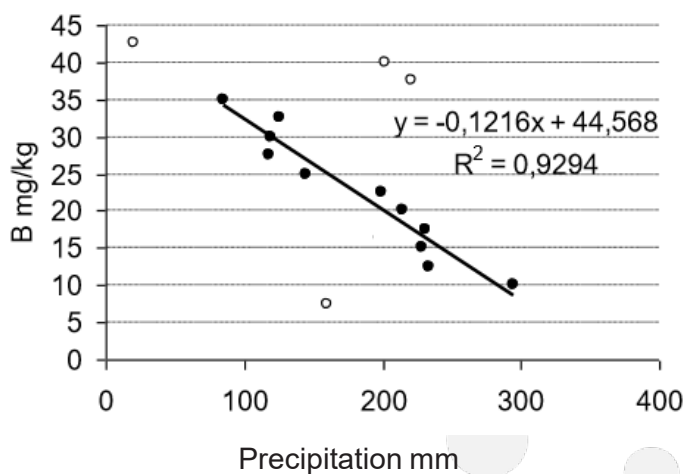
3- Strong precipitation events



Boron in Palm Oil

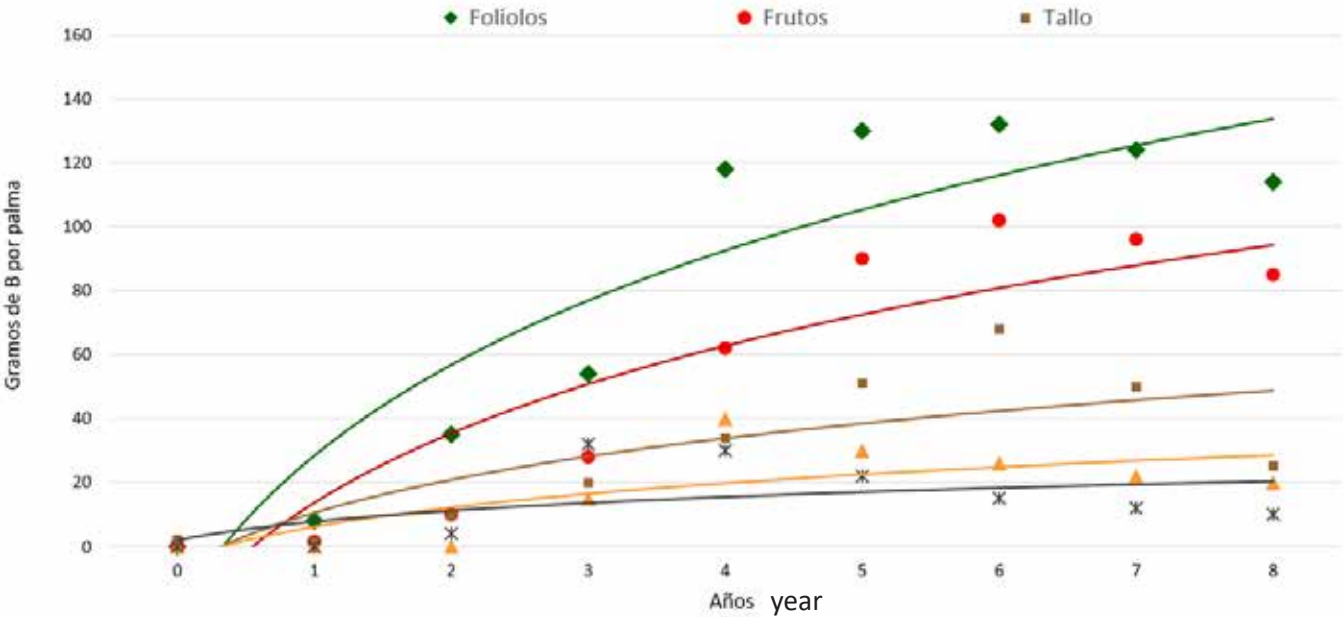
Strong Events of Precipitation and Boron

- dilute the B concentration of the soil solution
- B leaches from the zone of radical activity
- restitution from MO and other soil minerals is not fast
- transitory deficit of 1-2 days is already reflected in deformations when the sheet expands (Broshat 2005)



Boron in Palm Oil

Average Boron Demand in Palm Oil



Fuente: The Role of Boron in the Oil-palm (*Elaeis guineensis*) J. A. RAJARATNAM and J. B. LOWRY. Annals of Botany. New Series, Vol. 38, No. 154 (January 1974), pp. 193-200

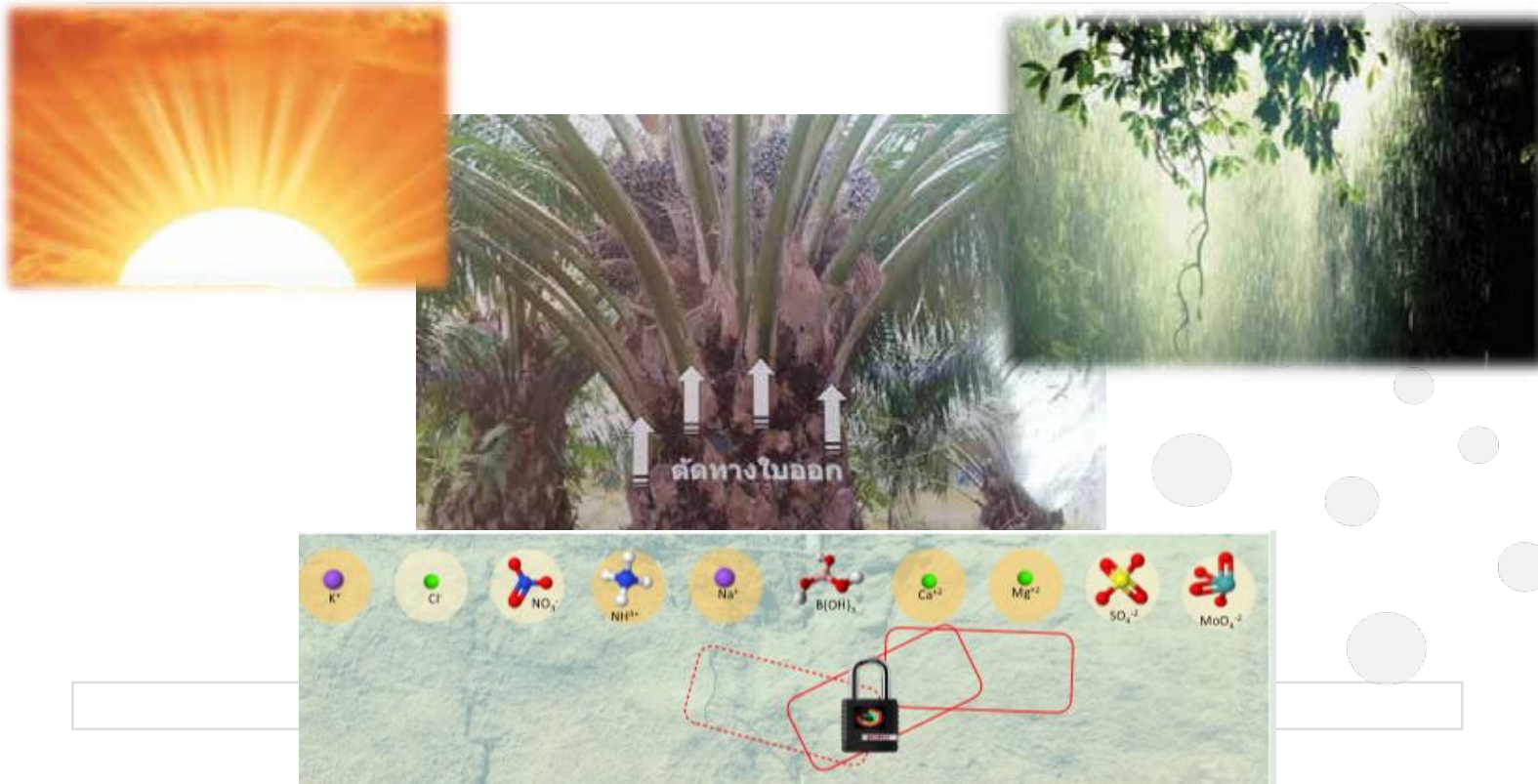
Foliolo=Leaf , Frutos Leaf Tallos=Steam

Symptoms of B deficiency and toxicity can be found in the same leaf



symptoms originate at different times of the leaf age

Bioavailability of Soil Boron



Application of a Boron Complex Foliar



Spraying Boron in Palm Oil

Metalosate Boron



Aminoacid Complex Boron

What is ? Metalosate

- **A specific chelate mineral fertilizer designed to satisfy and improve nutrients in plants**
- **The chelates used by Metalosate are Amino Acid the soy bean, where amino acids are the main ingredient for the production of proteins**





Research plantations Indonesia



PT. Peniti Sungai Purun

Metalosate Boron Application



0 Day



120 Day

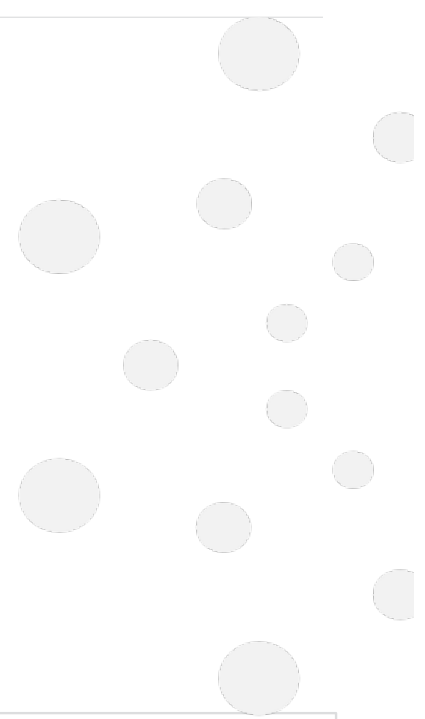
Spraying Metalosate Boron

Metalosate Boron Application









Spraying Metalosate Boron

Metalosate Boron Application



Spraying Metalosate Boron

Applications Metalosate Boron

Time	Treatment		
	Metalosate Boron, 1 application interval 6 month	Metalosate Boron , 1 applicata interval 12 month	Control (Boron Granular)
0 Day			
90 Days			

Spraying Metalosate Boron

Metalosate Boron Application



0 Day Metalosate Boron



180 Days Metalosate Boron

Spraying Metalosate Boron

Applications Metalosate Boron

Application Boron



0 Dias



90 Dias



Application
Metalosate
Boron



0 Dias



90 Dias



Application
control
Boron
granular soil

Spraying Metalosate Boron

60 day After Application



Metalosate Boron 1 l/ha



Boron Granular 80 gr/ plant

Honduras Central America

Application Metalosate

Fumigation equipment	Metalosate*	Doses	Lt Water	Interval ha	Application
sprayer pump	• Boron	1 l/Ha	70 lt	180 day	Spread uniformly on one side of the plant by one hectare leaves (1 sprayed plant ± 0.5 L)
Drone sprayer	• Boron	0.5 l/ha	30 lt	Cada 180 dias	Pulverized on leaves of plants for one hectare (1 sprayed plant ± 0,2 L).

Note * The type of Metalosate used is adapted to the condition of the plant

Metalosate Boron

Absorption Time of Different Formulation of Foliar Fertilizers (50% Absorption)			
Minerals	Salts and oxides	Synthetic Chelates	METALOSATE® FOLIAR FERTILIZER
Nitrogen (N)	1-6 hours	1-6 hours	Less than 12 minutes
Phosphorus (P)	15 days	7 to 11 days	Less than 2 hours
Potassium (K)	4 days	2 days	Less than 1 hour
Calcium (Ca)	6 days	3 days	Less than 2 hours
Magnesium (Mg)	5 hours	1 hour	Less than 1 hour
Sulfur (S)	12 days	8 days	Less than 2 hours
Iron (Fe)	2 days	1 day	Less than 2 hours
Manganese (Mn)	2 days	1 day	Less than 3 hours
Boron (B)	2 days	-	Less than 2 hours
Zinc (Zn)	3 days	1 day and 2 hours	Less than 2 hours

References:
 Graff, D. (1990). Stability constants of bivalent metals chelated into HVP and absorption there from
 Johnson, B. (1989). Physical characteristics of ALBION Chelates versus other chelates.

Metalosate Boron

Metalosate



**Metalosate: Micronutrients
chelated by amino acids
designed for foliar
application to crops**

Metalosate Boron



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PLANT NUTRITION

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